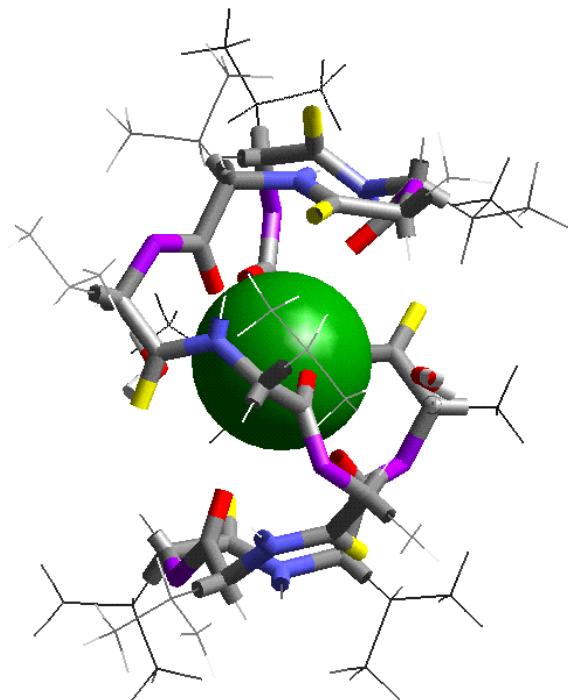


Organické sloučeniny kyslíku

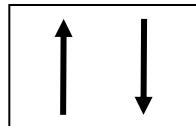


Matej Kohutiar

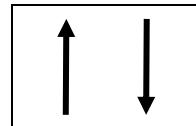
Organické sloučeniny kyslíku

- Alkoholy a fenoly
- Karbonylové sloučeniny
- Karboxylové kyseliny a deriváty
- Etherní sloučeniny

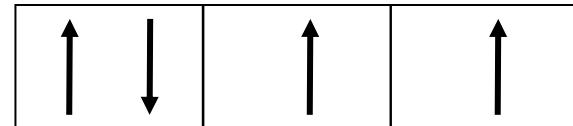
Elektronová konfigurace kyslíku



$1s^2$

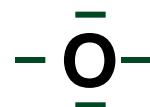


$2s^2$



$2p^4$

In organic molecule oxygen is attached covalently



Elektronové efekty

indukční efekt



Hydroxyderiváty uhlovodíků

- Alkoholy (C_{sp^3} -OH)
- Fenoly (C_{ar} -OH)
- Ethery (R-O-R)

Alkoholy



Alkoholy

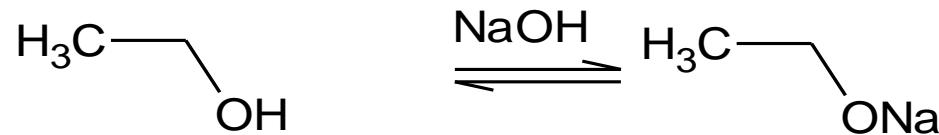
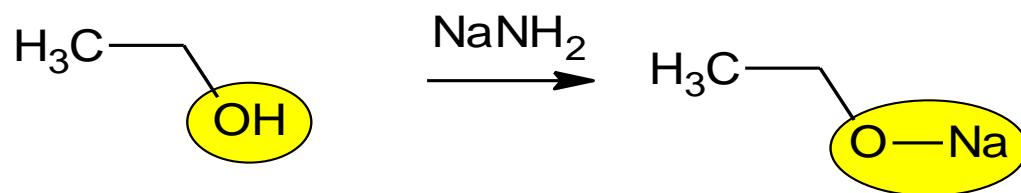
- 1° alkohol: 1 uhlík ve vazbě na uhlík s hydroxylovou skupinou
- 2° alkohol: 2 uhlíky ve vazbě na uhlík s hydroxylovou skupinou
- 3° alkohol: 3 uhlíky ve vazbě na uhlík s hydroxylovou skupinou

Alkoholy

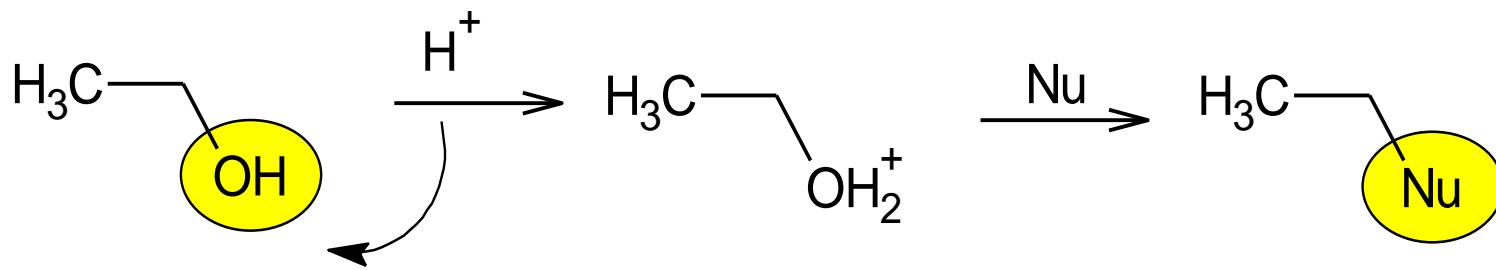
Podle počtu hydroxylových skupin v molekule:

- Monohydroxyderiváty
- Polyhydroxy alkoholy (polyoly)
 - Dioly (dihydroxyderiváty)
 - Trioly (trihydroxyderiváty)
 - Tetroly (tetrahydroxyderiváty)
- Fenoly: -OH ve vazbě na aromatický cyklus

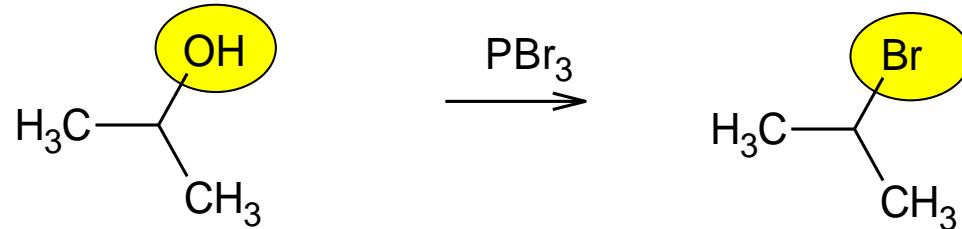
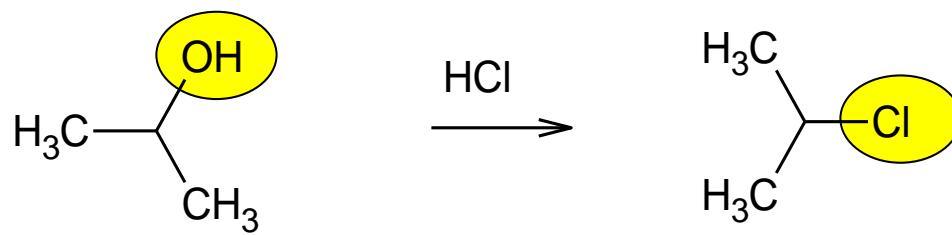
Acidita a alkalita



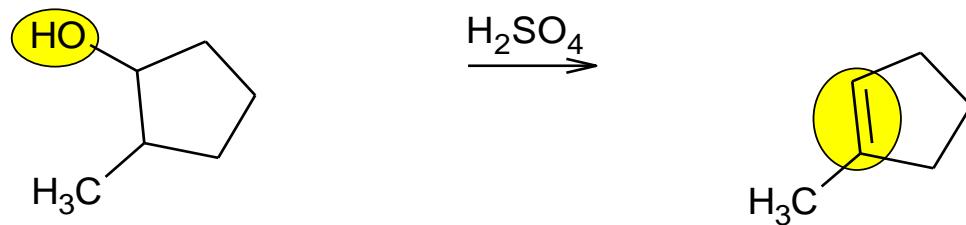
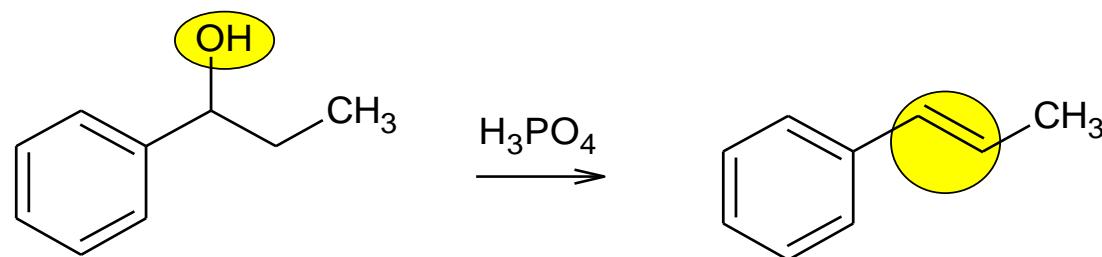
Nukleofilní substituce



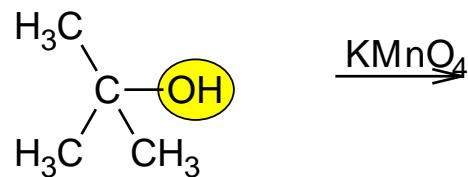
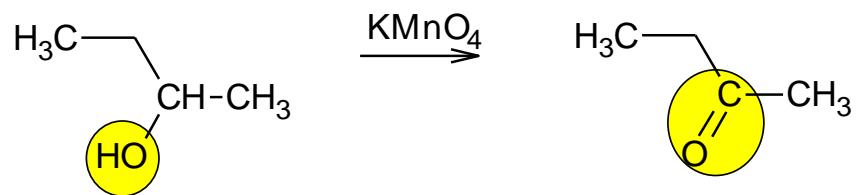
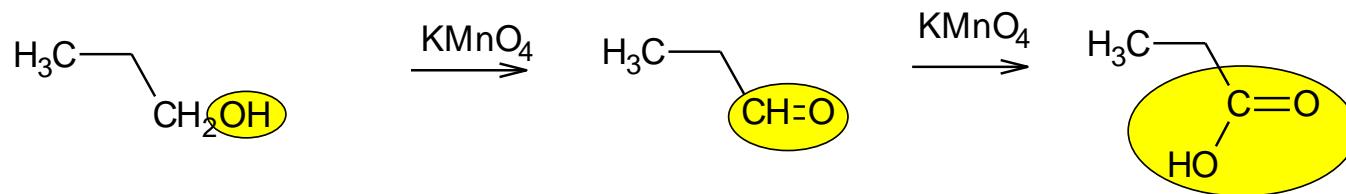
Nukleofilní substituce



Eliminace



Oxidace alkoholů

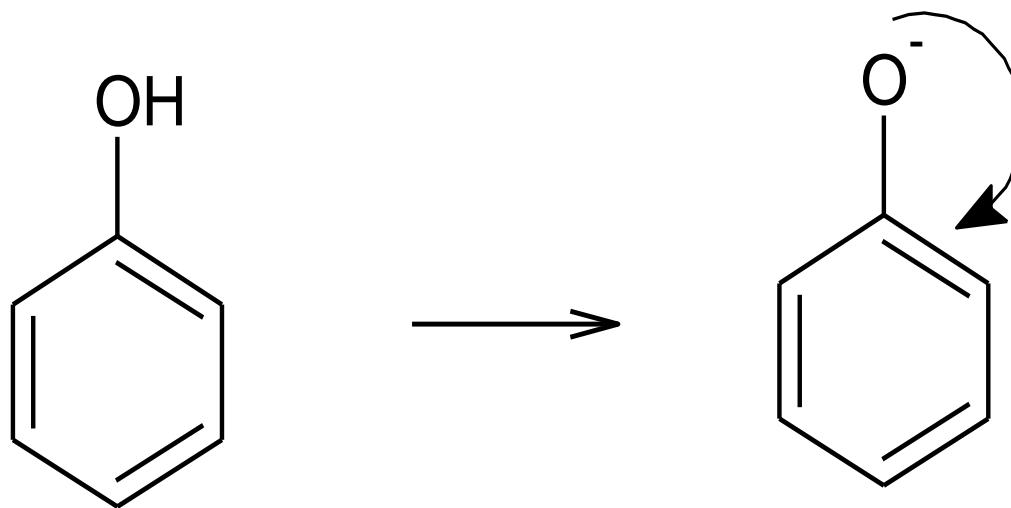


Biologické vlastnosti alkoholů

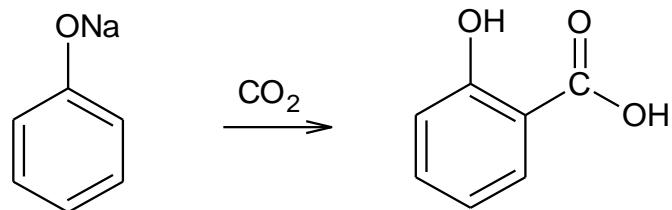
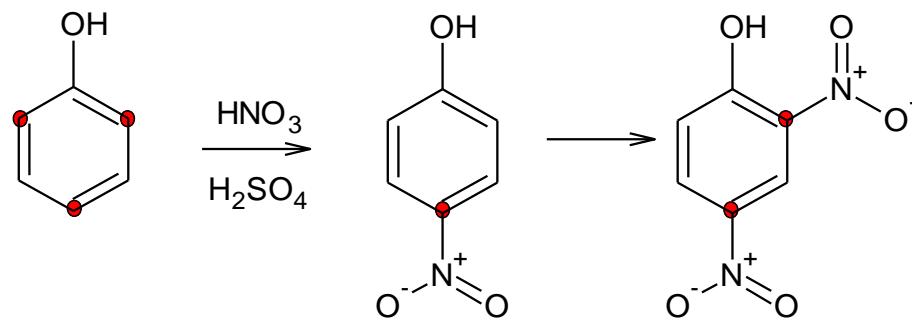
- Molekuly léčiv často obsahují OH skupinu
 - Zvýšení polarity a rozpustnosti ve vodě
 - Hypnotická aktivita stupá od terciárních k primárním a.
 - Steroidy, hormony
-
- Metanol
 - Etanol
 - Glycerol

FENOLY

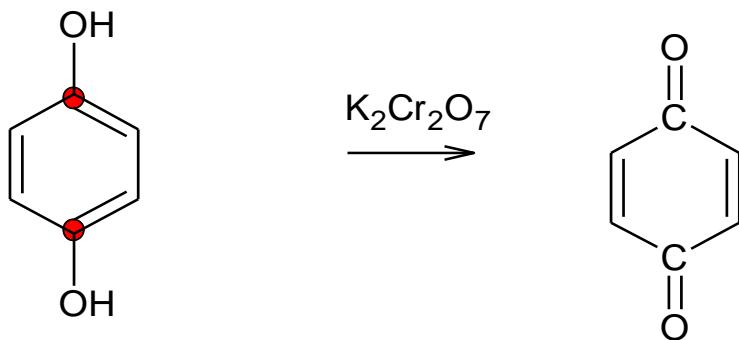
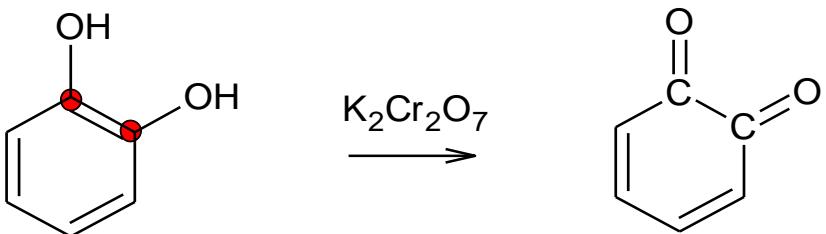
Fenoly



Reakce fenolů



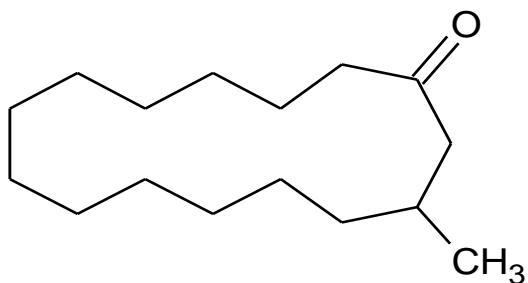
Reakce fenolů



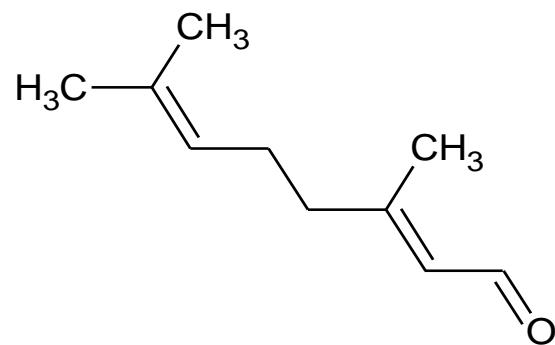
Biologické vlastnosti fenolů

- Velmi reaktivní skupina
- Fenol
- Resorcinol
- Salicylová kyselina
- Acetylsalicylová kyselina

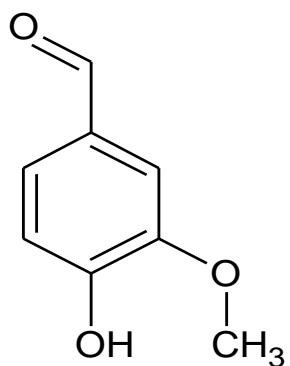
KARBONYLOVÉ SLOUČENINY



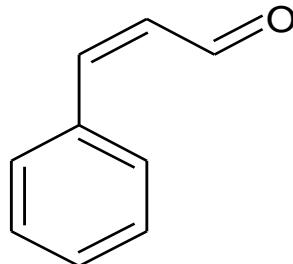
muscone



citral

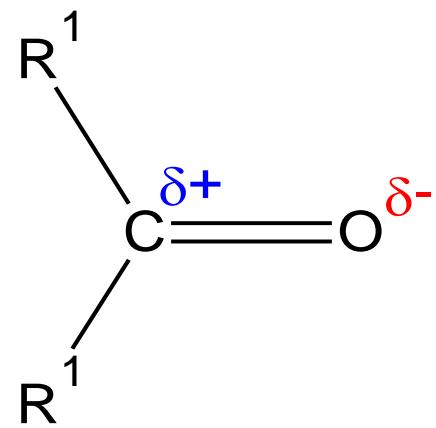


vanillin

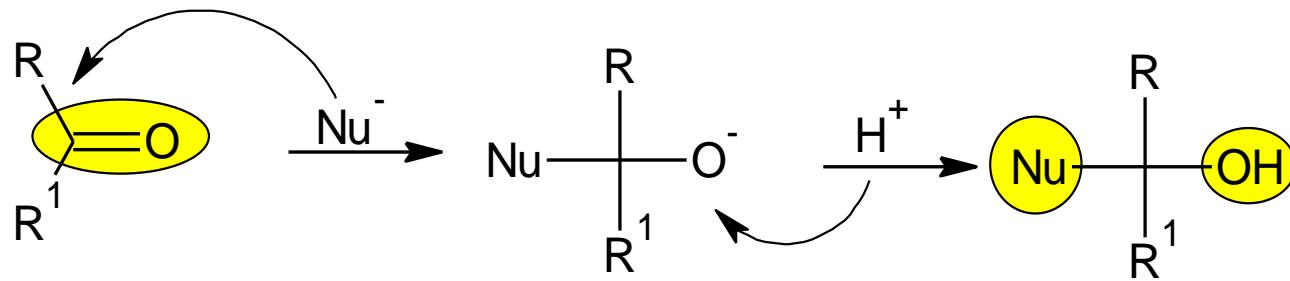


cinnamaldehyde

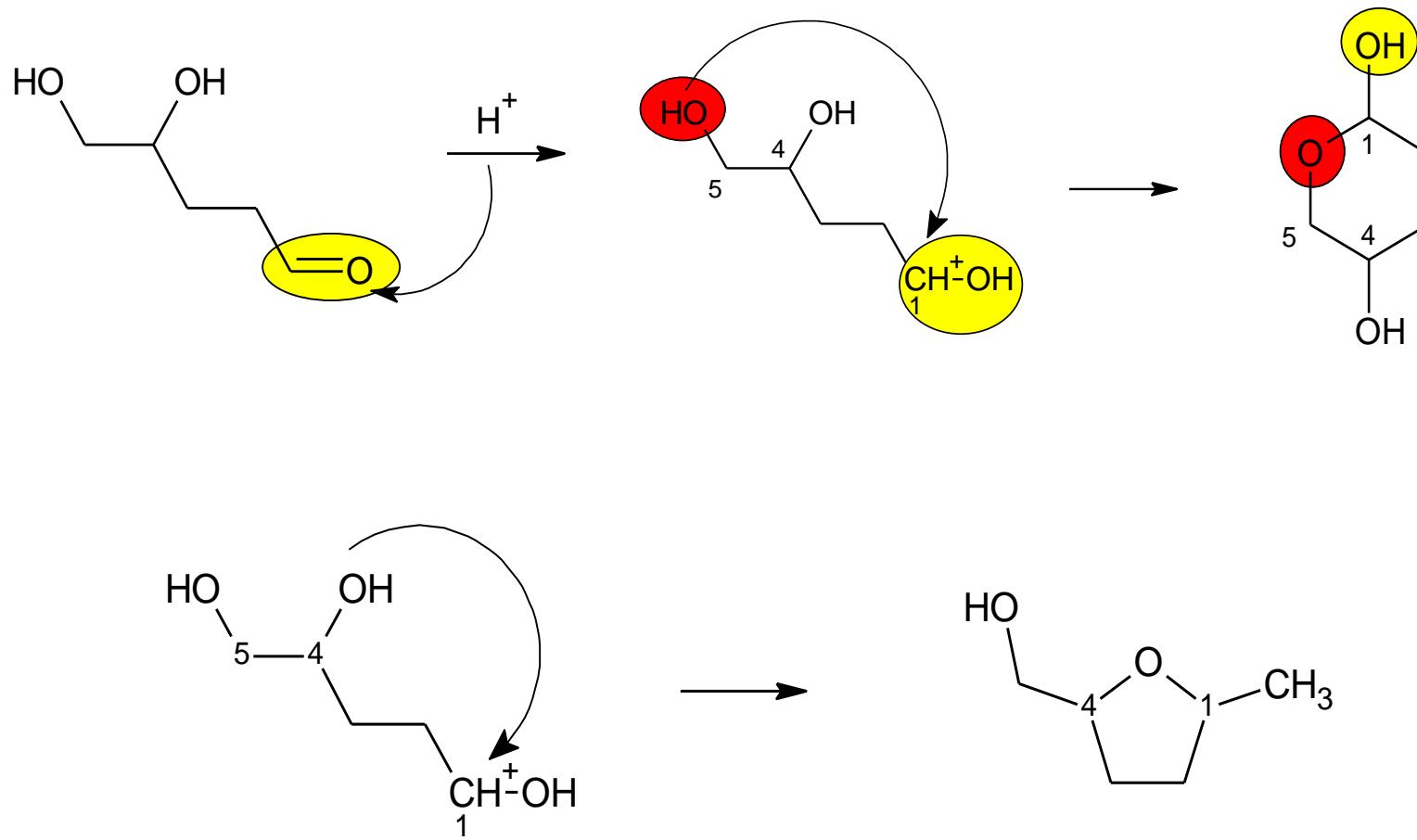
Reaktivita



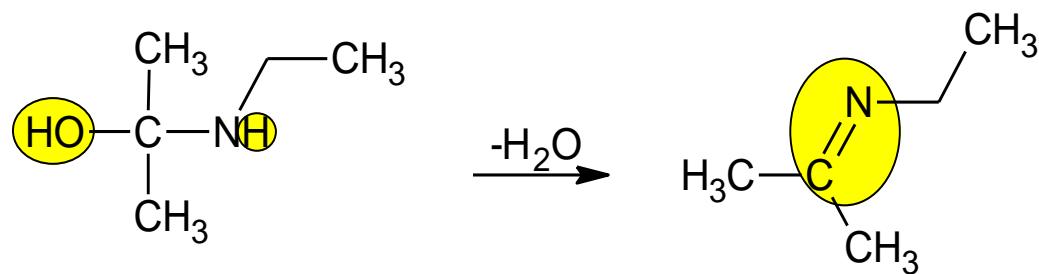
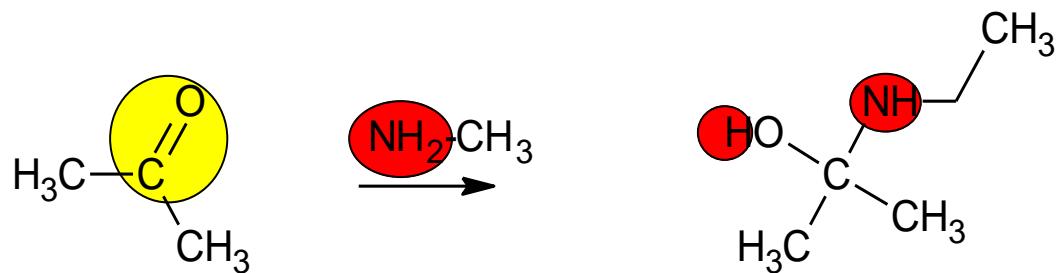
Nukleofilní adice



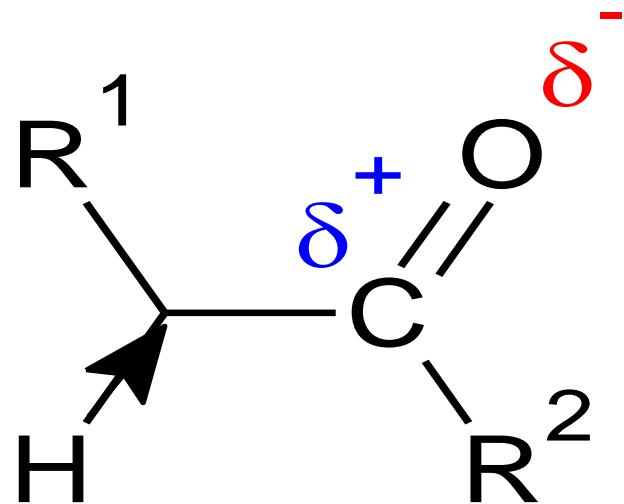
Nukleofilní adice



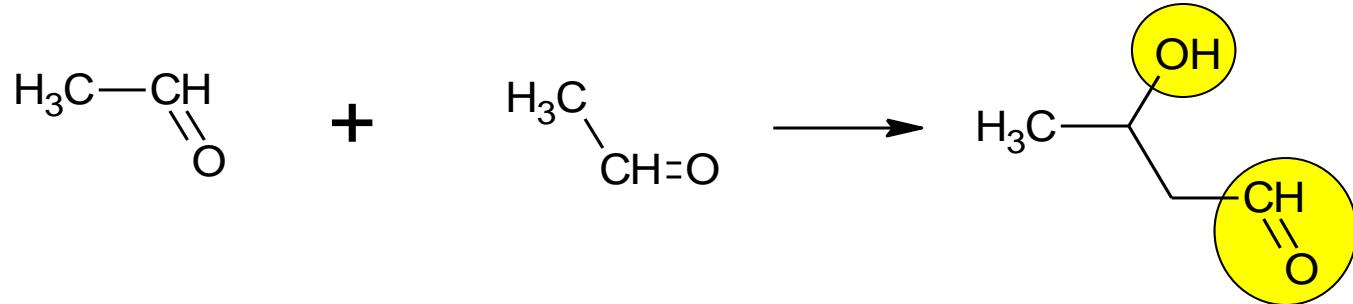
Nukleofilní adice



Aldolová kondenzace

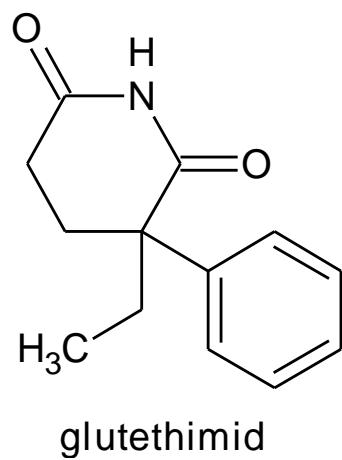
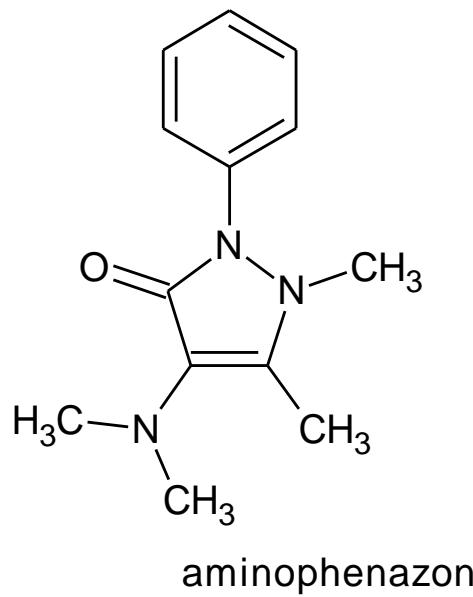
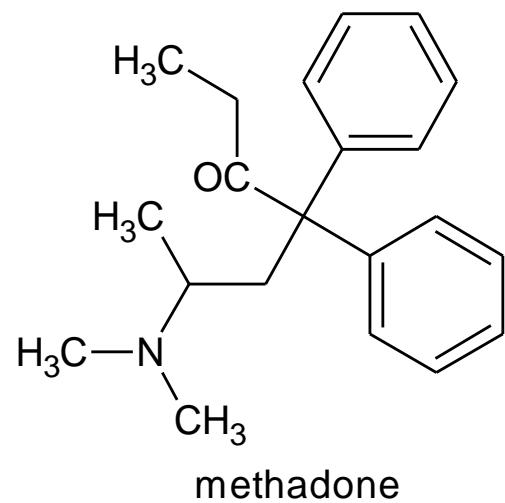


Aldolová kondenzace



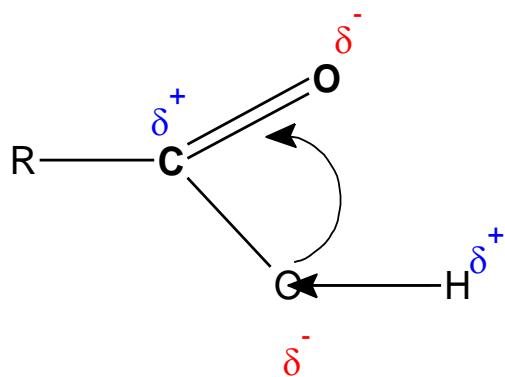
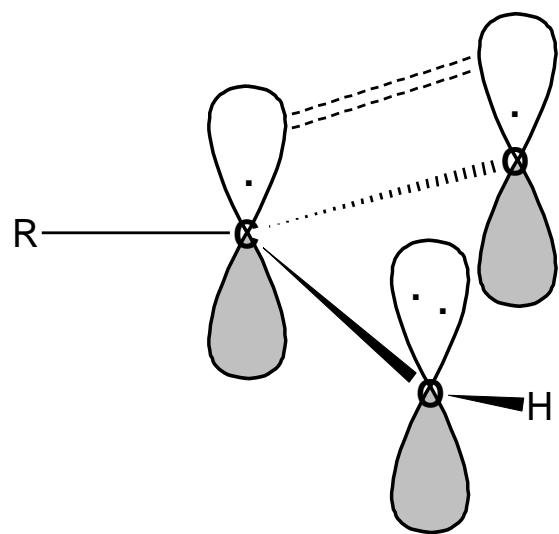
Biologické efekty karbonylových sloučenin

- Reaktivní skupina
- Formaldehyd
- Benzaldehyd
- Aceton
- Steroidní hormony

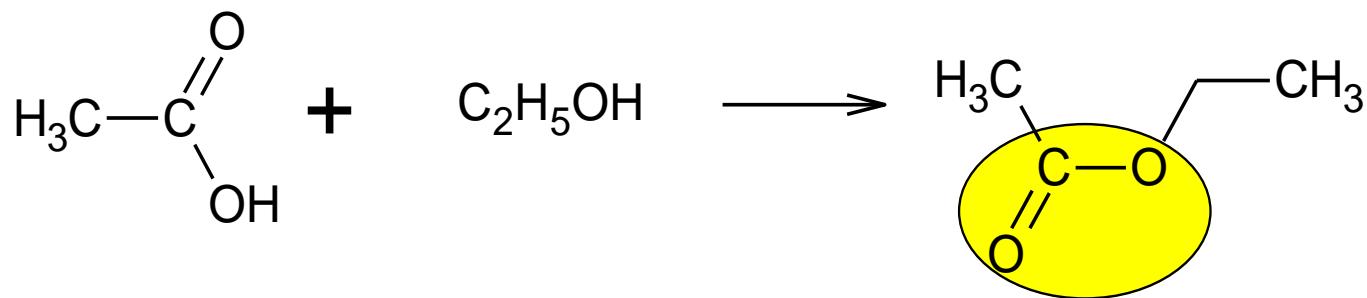


KARBOXYLOVÉ KYSELINY

Karboxylové kyseliny

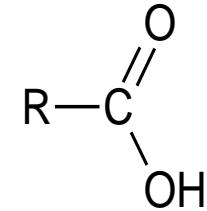
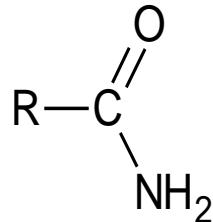
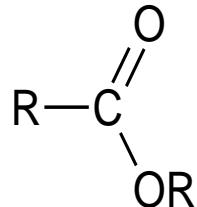
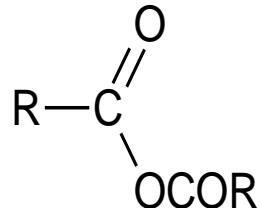
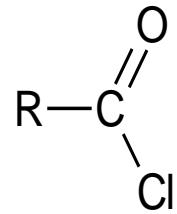
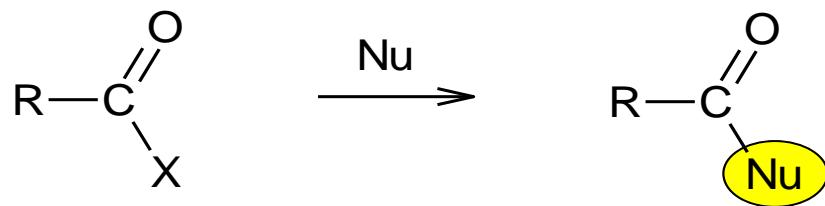


Nukleofilní substituce *esterifikace*

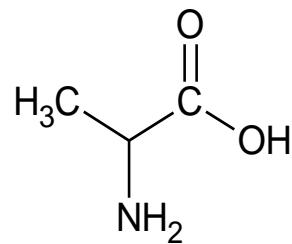
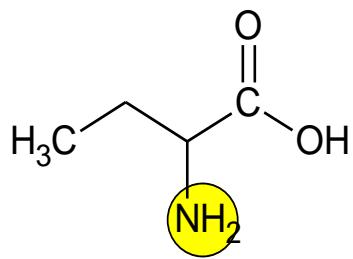
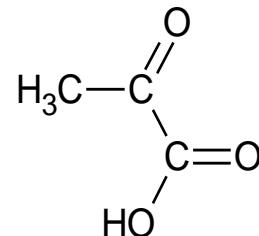
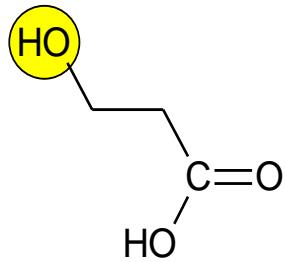
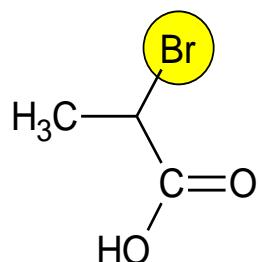


DERIVÁTY KK

Funkční deriváty

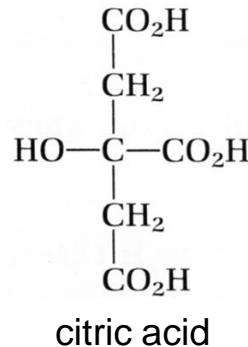
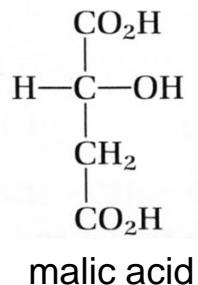
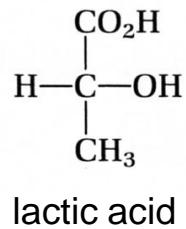
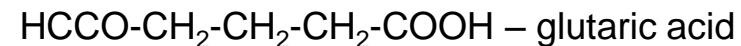
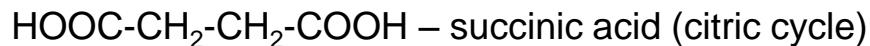


Substituční deriváty



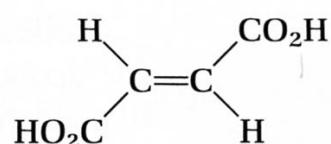
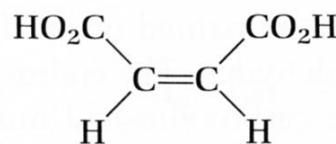
Důležité substituční deriváty KK

Dicarboxylic acids



-OH group containing acids

Ketoacids



Maleic acid and fumaric acid are geometric isomers

